



62nd NDIA Annual Fuze Conference
Buffalo, NY
May 13-15, 2019

Innovative Developments in Carbon Cathode Matrix Materials for Li/SOCl₂ Reserve Batteries

Brett Barclay
Design Engineer
EnerSys Advanced Systems
Horsham, Pennsylvania
(215) 773-5487

Juan Munoz
Design Engineer
EnerSys Advanced Systems
Horsham, Pennsylvania
(215) 773-5487

This presentation includes forward-looking statements and/or information, which are based on the Company's current expectations and assumptions, and are subject to a number of risks and uncertainties that could cause actual results to materially differ from those anticipated. Such risks include, among others, risks associated with competitive actions, technology development and implementation, intellectual property infringement, failure to execute acquisition strategy, penetration of existing markets, expansion into new markets, hiring and retaining high quality management and key employees and general economic conditions including the risks described in the Company's most recent annual report on Form 10-K dated and filed with the SEC on [May 30, 2018](#) along with other unforeseen risks. The statements in this presentation and accompanying commentary are made as of the date of this presentation, even if subsequently made available by the Company on its website or otherwise. The Company does not undertake any obligation to update or revise these statements to reflect events or circumstances occurring after the date of this presentation

- Li/SOCl₂ Reserve Battery Applications & Characteristics
- Li/SOCl₂ Electrochemistry
- Role of Carbon Matrix Material
- Development of New Material
- Mechanical Strength
- Porosity Metrics
- Electrochemical Performance
- Process and Formulation Tradeoffs

Lithium oxyhalide battery applications include

- Artillery
- Air delivered Weapons
- Mortar Munitions
- Missiles/Rockets
- Barrier Munitions

- Long shelf life: > 20 years
- Ability to be fabricated in a variety of configurations
- Large usable temperature range (-60°C to 85°C)
- Open circuit voltage of 3.7 V per cell
- Flat discharge profile
- High energy density
- Moderate rate capability (typical maximum current density of 50 mA/cm²)

Lithium: Solid metal anode (where oxidation occurs)

Thionyl Chloride: Liquid cathode serving as its own solvent (catholyte) (where reduction occurs)

Anode Reaction:



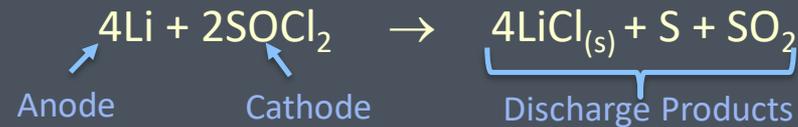
Cathode Reaction (at carbon surface):



Overall reaction:



Overall reaction:

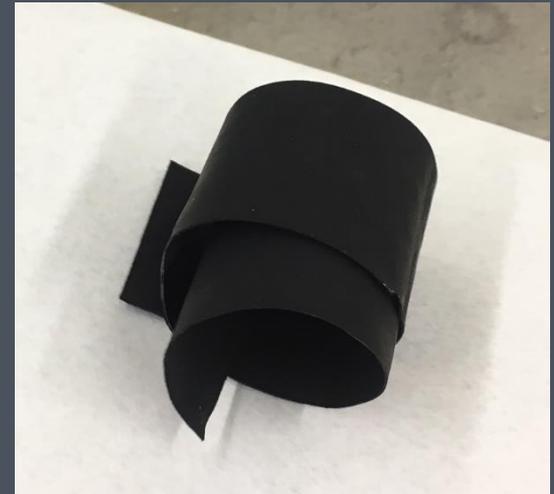


- Carbon supplies energetically favorable deposition sites for the discharge product: LiCl
- Typically referred to as the cathode matrix material even though it does not directly contribute any electrochemical potential or energy to the system
- The carbon presence and specific properties influence the cell's realized energy and rate capability – it is almost always the limiting reagent
- This occurs due to its role as a facilitator in the discharge reaction at the carbon surface
- Made up of carbon and PTFE. The PTFE acts as a binder.

Carbon properties which facilitate the maximum amount of deposition sites for the LiCl discharge product, while maximizing mass and charge transfer rates will result in a cell with optimal electrochemical performance

These metrics consist of:

- Porosity
- Pore volume
- Pore diameter
- Tortuosity
- Surface area
- Density
- Conductivity
- Overall Formulation/'Active' Material



Sample of carbon matrix material
produced at EAS

Goals

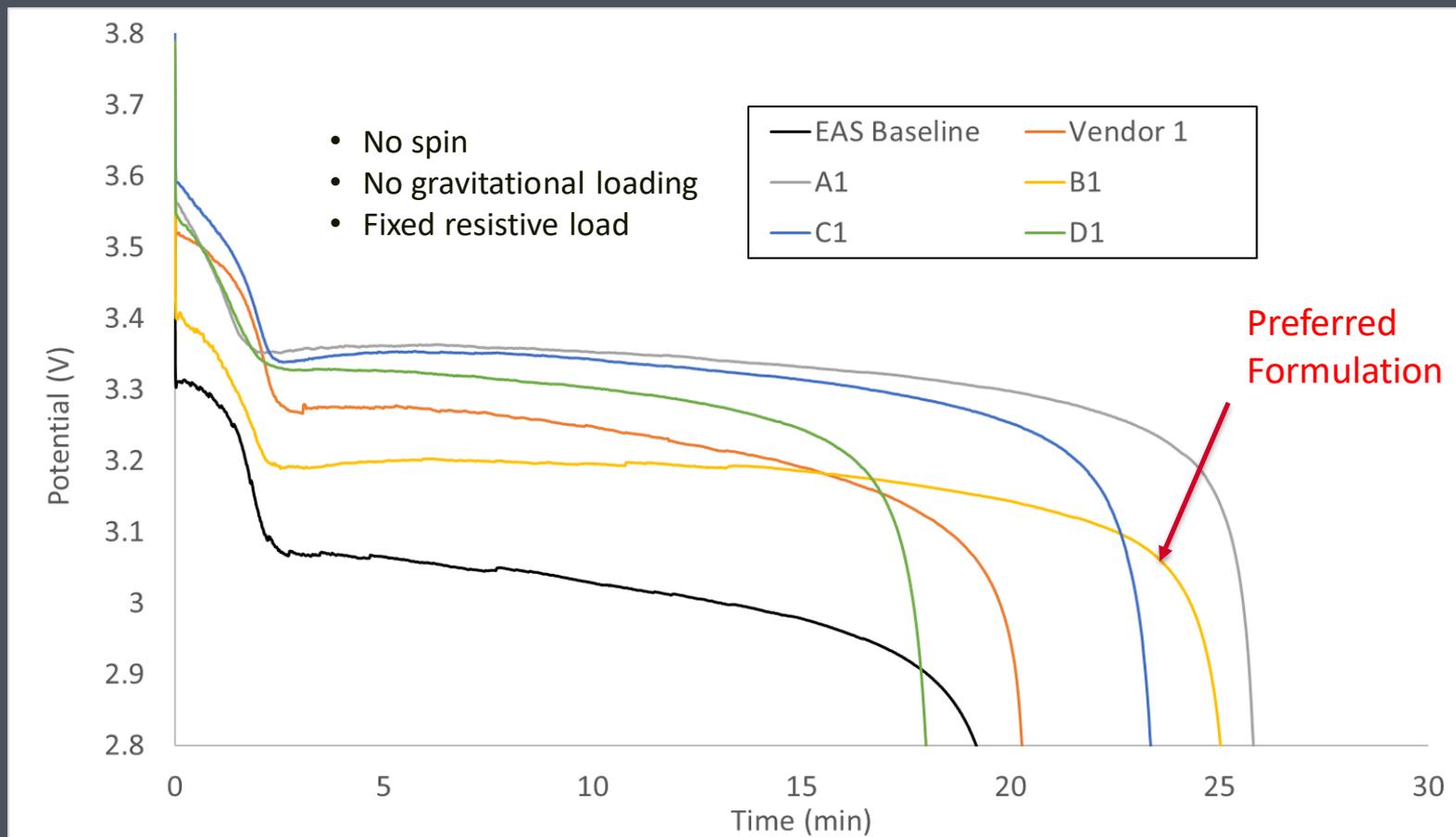
- Improve electrochemical performance
- Produce production-friendly material in both handling and production capabilities

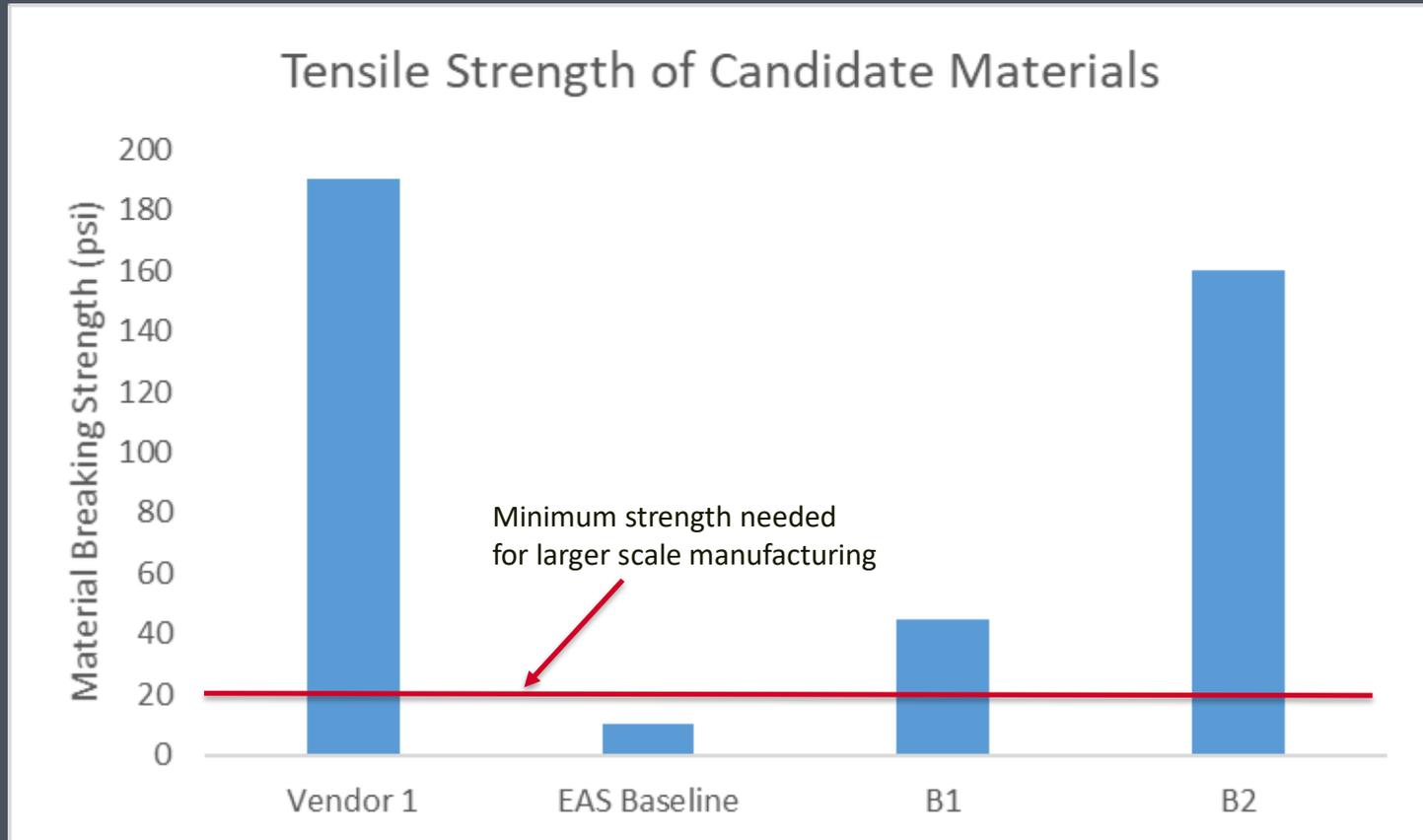
Theory – carbon matrix material consists of high surface area carbon and PTFE. The following can be varied to influence the mechanical and electrochemical performance metrics

- Carbon/PTFE ratio – Developed a process for 4 formulations (A, B, C, D)
- Handling of the material after introduction of the components (Method 1 or 2)

Sample Identification		
Carbon/PTFE Formulation	Method	
	1	2
A	A1	-
B	B1	B2
C	C1	-
D	D1	-

Initial evaluations performed in EAS' G2666B1 device at 63°C showed up to 25% improvement in run time to EAS Baseline and Vendor 1 material. using no spin or gravitational loading.





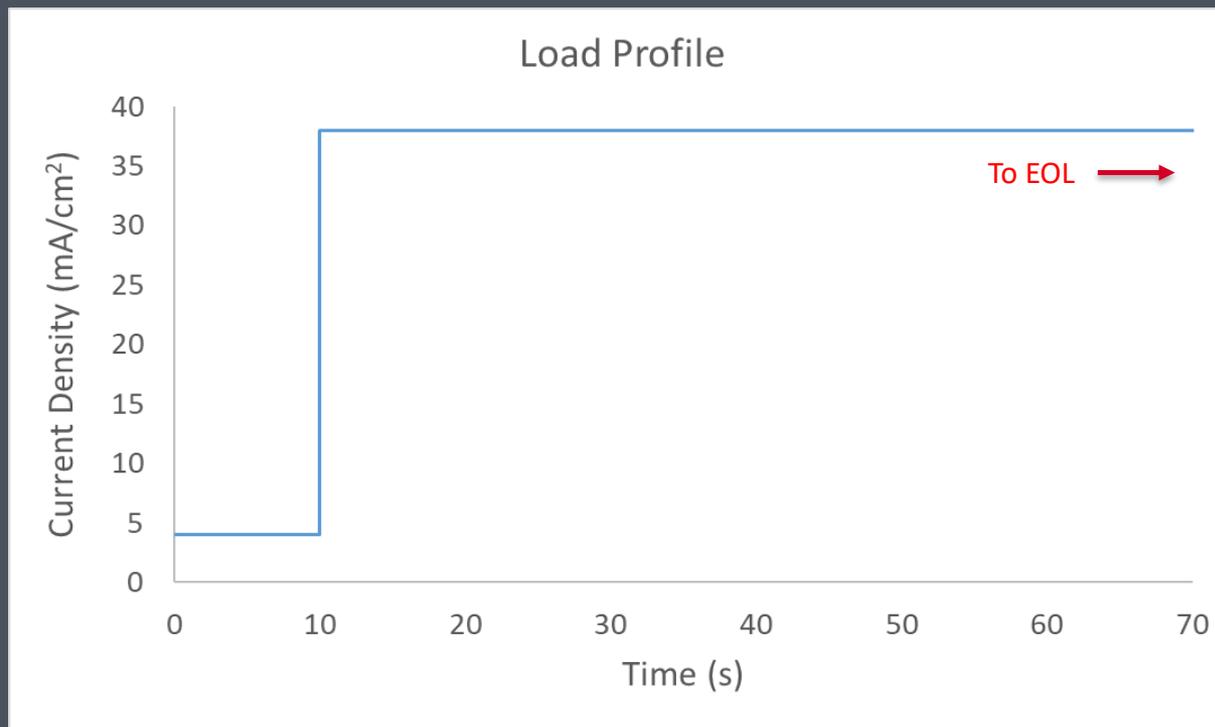
- Mechanical strength affects manufacturability, and handling characteristics
- Benefits and drawbacks exist for each material

Matrix Material	Porosity Metric				
	Surface Area (m ² /g)	Porosity (%)	Density (g/mL)	Pore Volume (mL/g)	Avg Pore Diameter (um)
Vendor 1	12	46	0.40	0.19	0.74
EAS Baseline	14	51	0.36	1.41	0.405
EAS B1	62	77	0.44	1.93	0.125
EAS B2	58	62	0.68	0.93	0.065

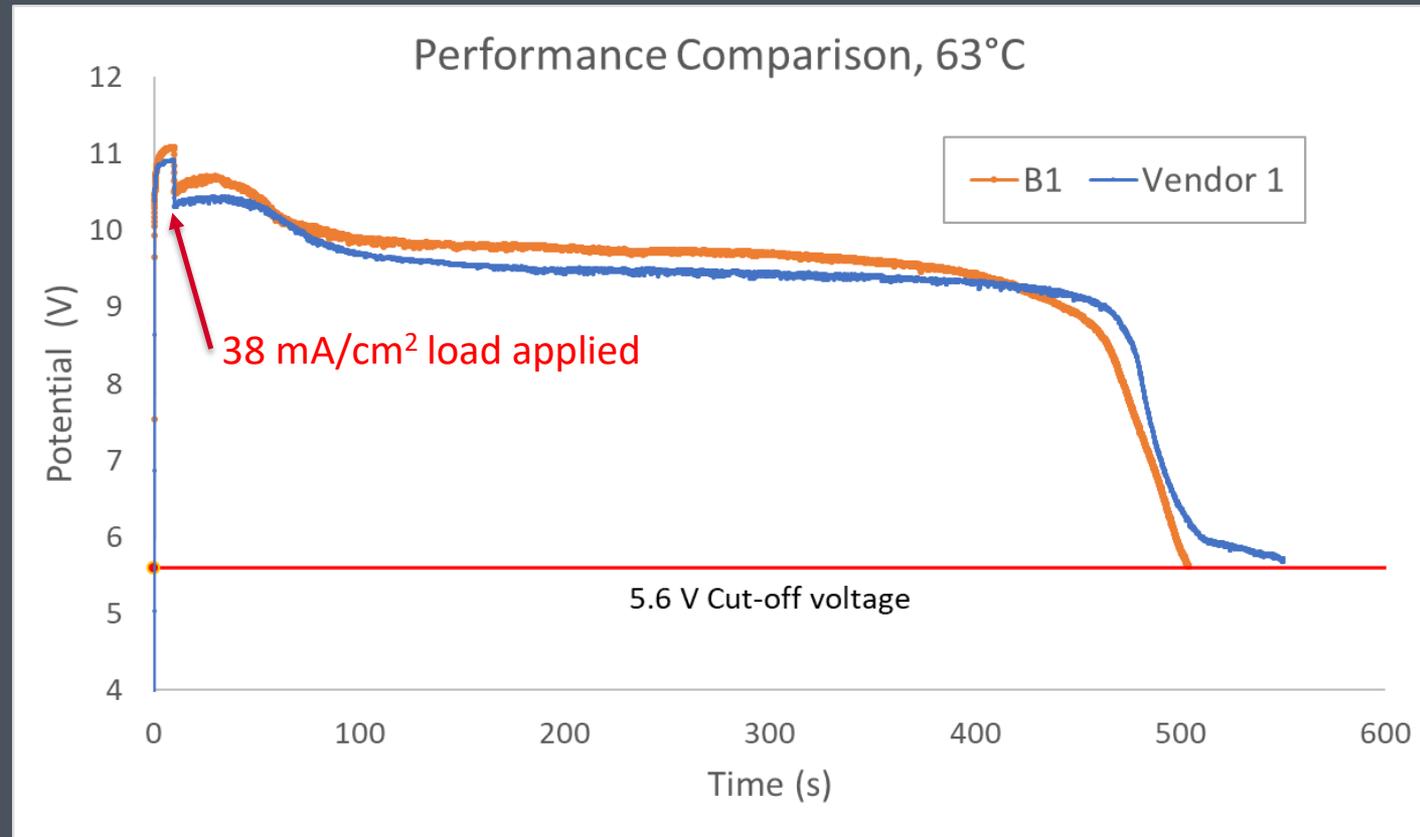
- Newly developed matrix material formulations exhibit significantly more favorable porosity metrics
- As long as mass transfer is not inhibited by small pore diameter or high tortuosity, we expect to see better electrochemical performance in the new formulations

Application-representative environment test setup:

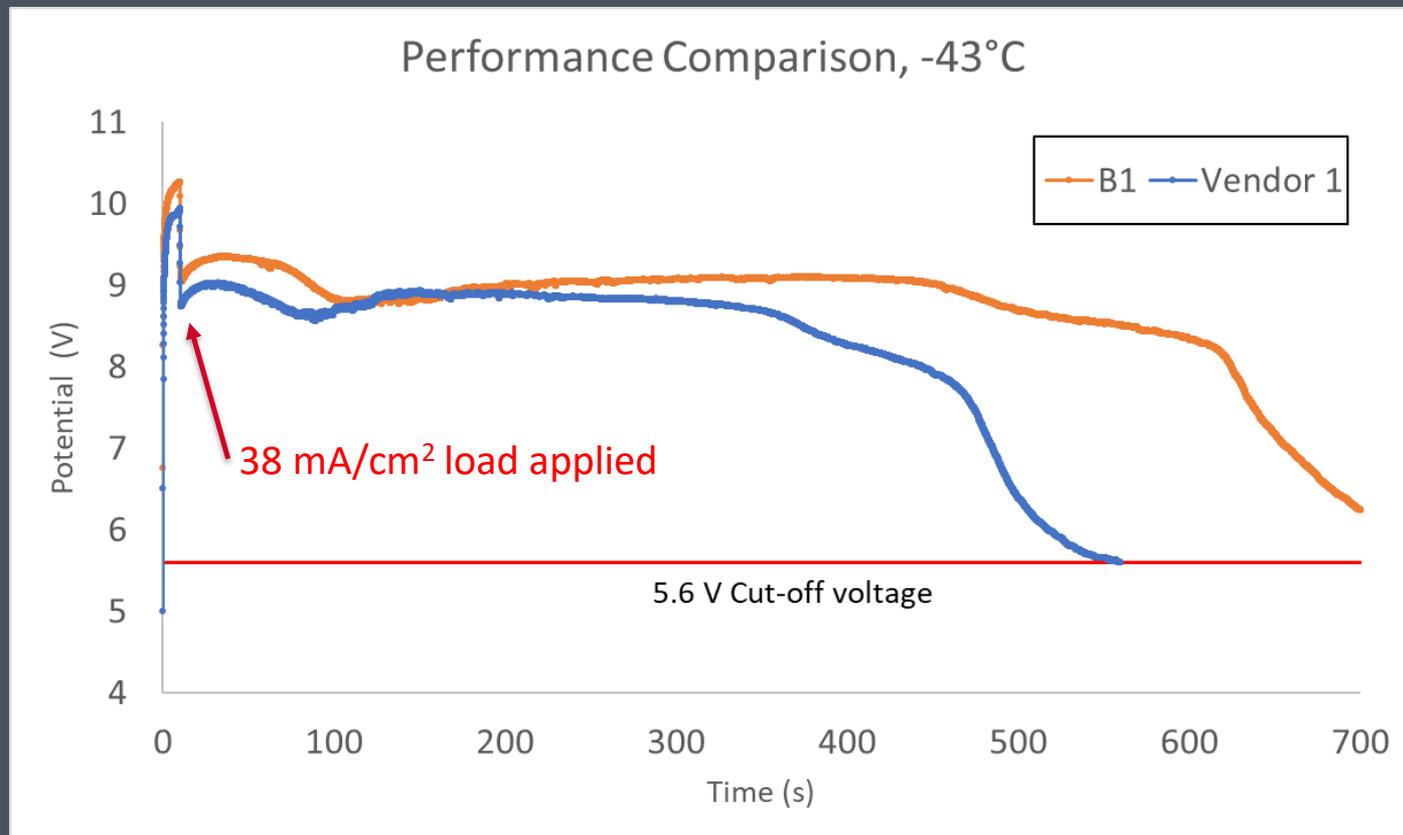
- Built into an existing EAS device with 3S2P configuration
- Tested in a spin-capable airgun at EAS
- Activated via gravitational loading
- Activated into 80 RPS spin
- Discharge rate of 38 mA/cm²



- Comparable run time seen at 65°C
- New formulation typically runs at slightly higher voltage

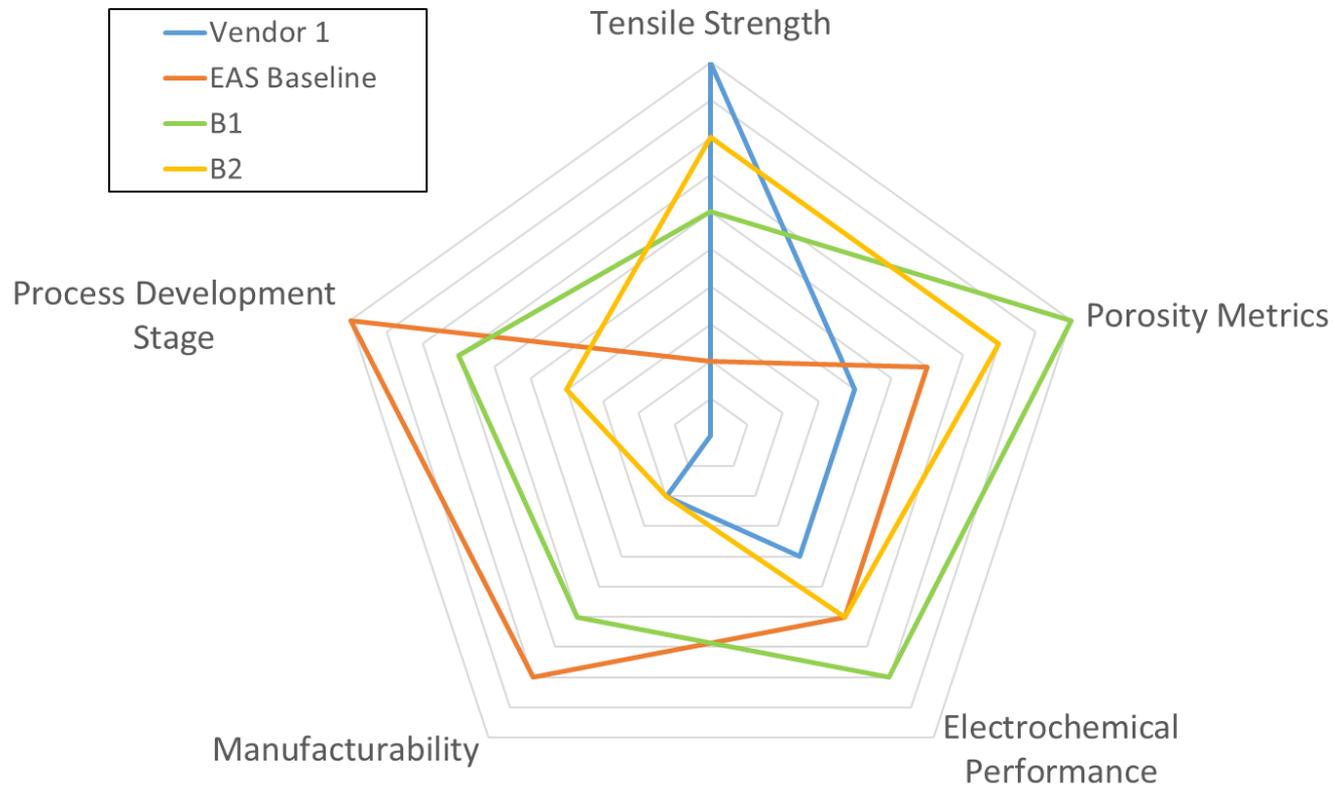


- New formulation runs ~35% longer at -43°C.
- New formulation typically runs at slightly higher voltage



- Vendor 1 material
 - Greatest tensile strength
 - Poor porosity metrics possibly influencing performance.
 - Risky for EAS to rely on a sole source with no control over it
- EAS Baseline
 - Low tensile strength, impacts ability to use at scale
 - Easy and relatively quick to manufacture
 - Serviceable electrochemical performance
- Formulation B1
 - Serviceable tensile strength for scalable production operations
 - Great porosity metrics
 - Great electrochemical performance, especially at cold temperatures
- Formulation B2
 - High tensile strength
 - Good porosity metrics
 - Not very scalable, lots of processing required

Ranking of Carbon Matrix Materials



- Successfully developed a new carbon cathode matrix material that is compatible with Li/SOCl_2 reserve battery chemistry
- Formulation B1 performs better electrochemically than baseline and commercially available material (likely due to porosity metrics), especially at cold temperatures
- The new process is able to create material with sufficient tensile strength to be used in higher-volume production (Formulation B1 and B2) and compete with commercially available material for electrochemical performance and supply chain reliability
- Future work will revolve around further characterization of Formulation B1 and scale up of the manufacturing process