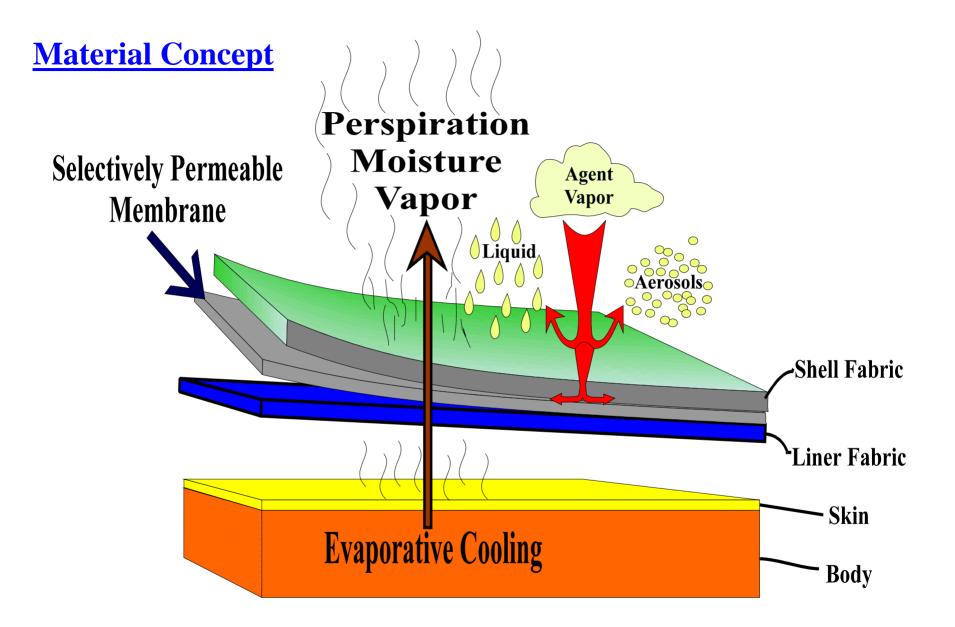


MEMBRANE DEVELOPMENT FOR THE NEXT GENERATION OF CHEMICAL BIOLOGICAL PROTECTIVE CLOTHING



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Vapor (V) Chemical Agent Protective Closure Systems











Vapor, Aerosol, Liquid, (VAL) Chemical, and Biological Agent Protective Closure Systems







All Purpose Personal Protective Ensemble (AP-PPE)

- Based on selectively permeable membranes
- Increased protection from liquids and aerosols
- Reduced weight and bulk
- Improved comfort and compatibility
- Improved operational suitability
- Reduced shelf-life burden

Membranes – Where do we go from here?

- •Optimize permselectivity
- •Ensure protection vs. toxic industrial chemicals (TICs)
- Introduce self-detoxification
- Integrate compatible closures

Ion Implanted Membranes

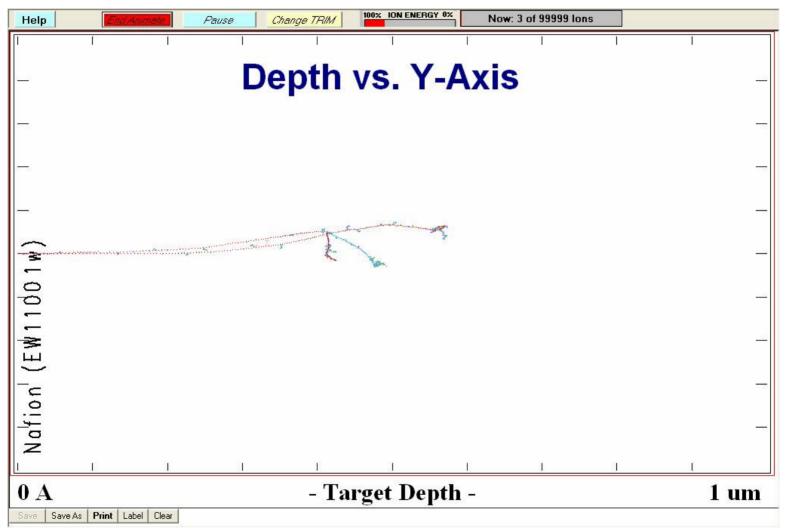
- Improve the permselectivity of membrane materials for use in chemical/biological (CB) protective clothing through the ion beam modification of the surface layers of available membranes
- Two-fold approach: computer modeling of the irradiation process to develop a better understanding of the process at the molecular level and irradiation experiments of materials at different energy levels and with different ions for permselectivity measurements.
- Correlation between the two efforts will ultimately yield a powerful tool for the development of permselective membranes for CB garments.

Nafion[®] (du Pont): $[-(CF_2CF_2)_n - (CF_2CF(OCF_2CF(CF_3)OCF_2CF_2SO_3H)) -]_x$

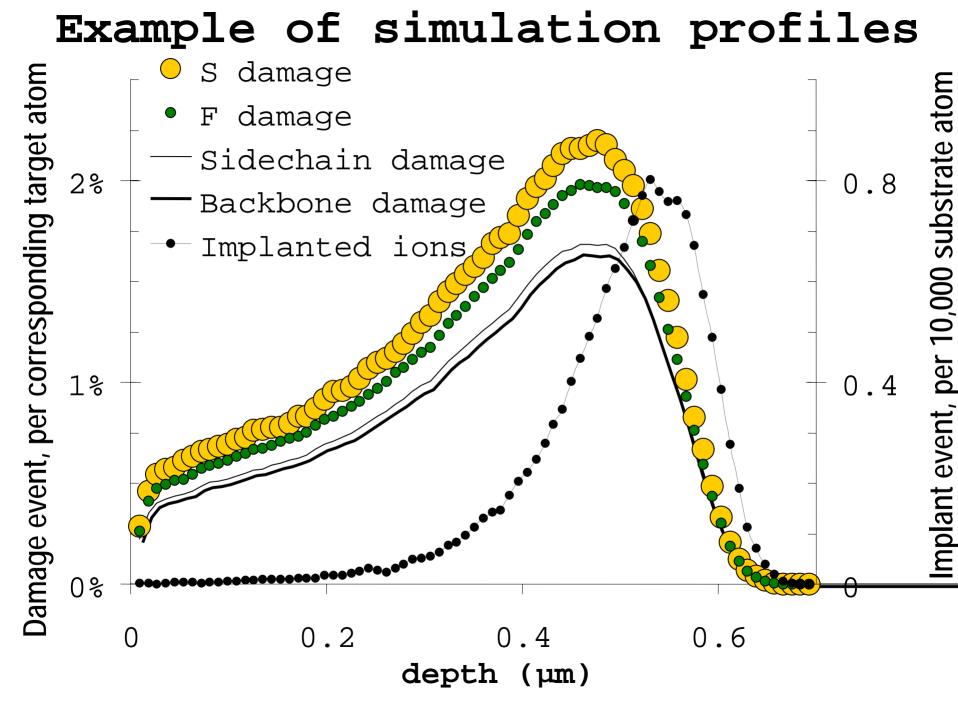
<mark>О-Н</mark> Н Na <mark>О-Н</mark> Н O-H 0-Н Н Na⁺ Na <u>О-Н</u> н О-Н Н Na

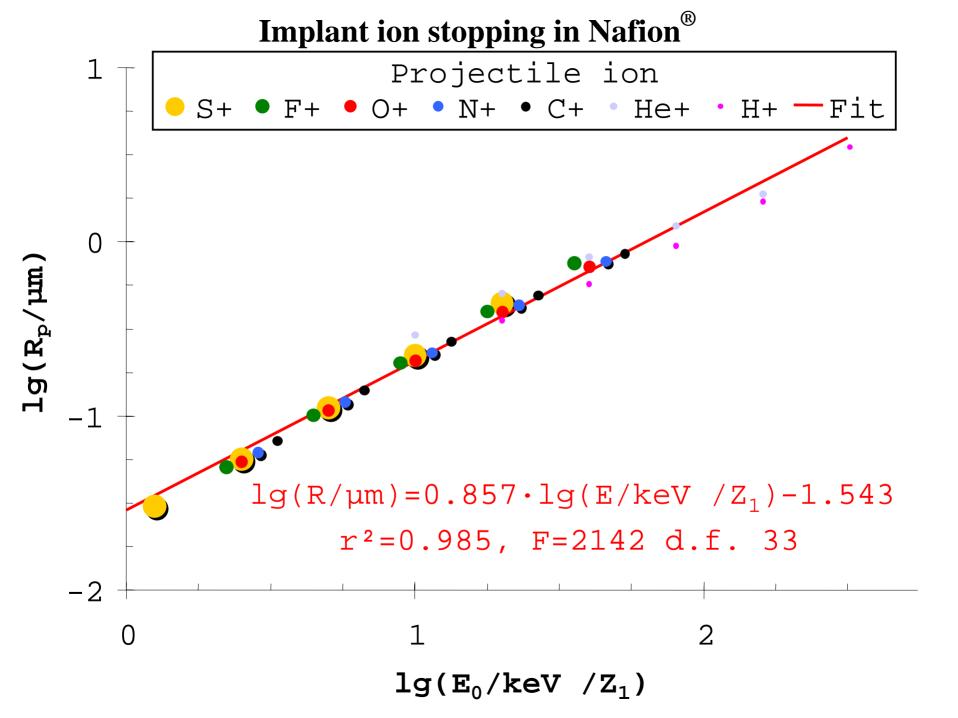
Schematic model of Nafion: hydrophilic, intermediate and hydrophobic phases

Transport of Ions in Matter (SRIM v.2000.41)¹

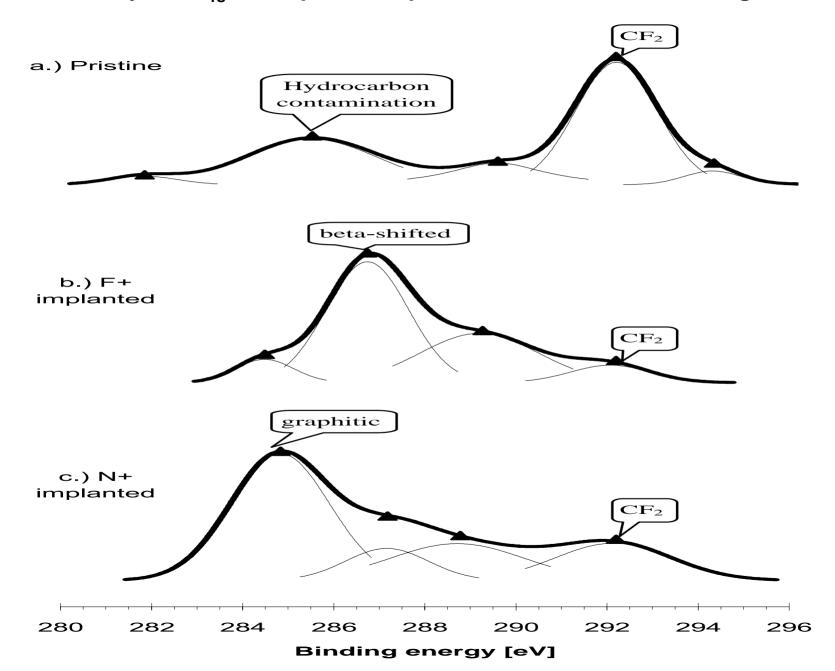


¹Ziegler, J.F.; Biersack, J.P.; and Littmark, U., The Stopping and Range of long in Solids (1985)

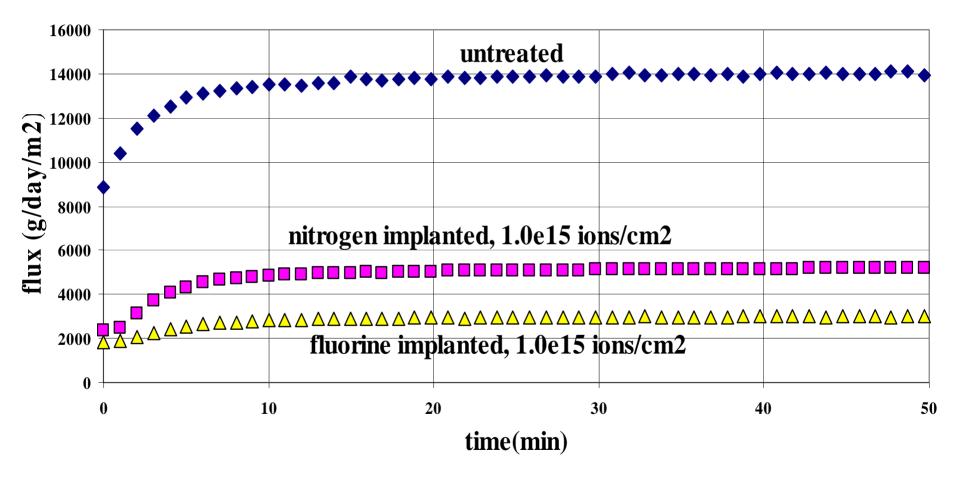




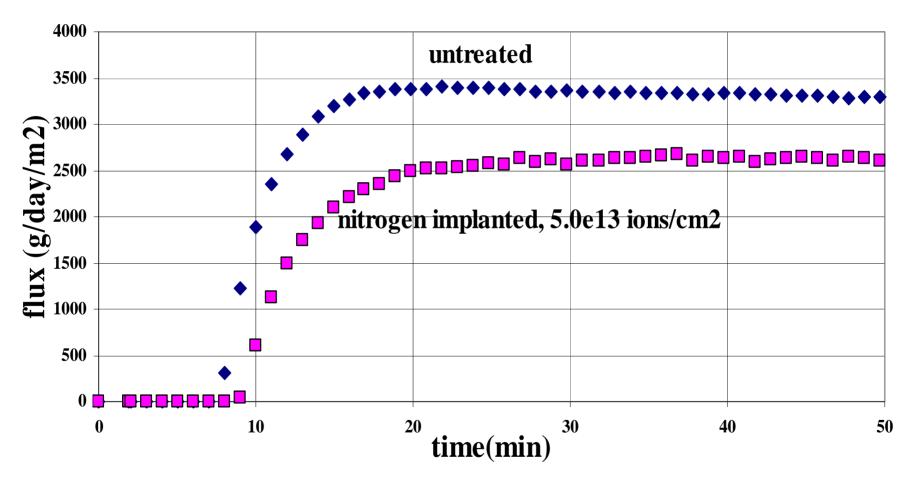
Surface analysis: C_{1s} XPS spectra of pristine and ion beam damaged Nafion®



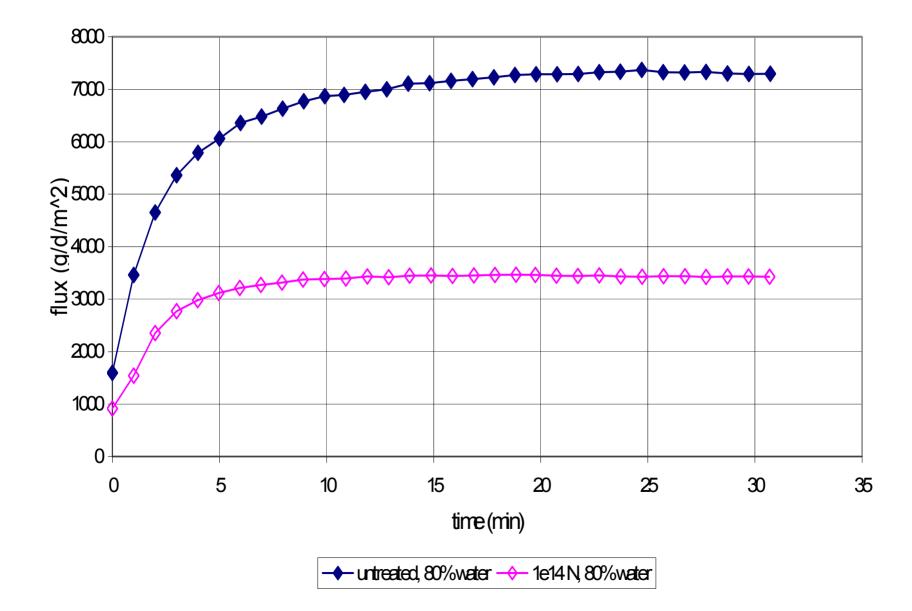
Water Permeation in NAFION 117



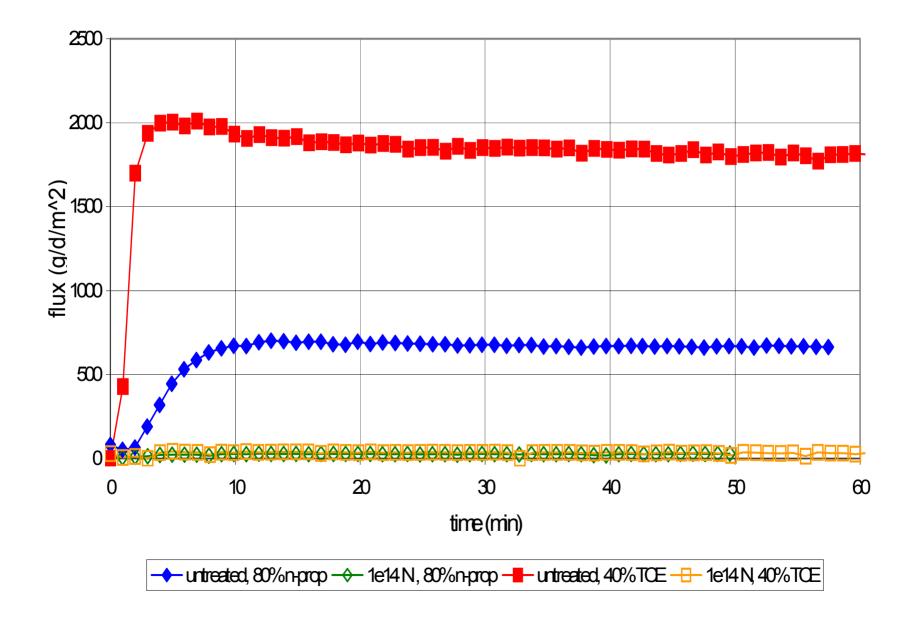
N-propanol Permeation in NAFION 117



Protolyte A700

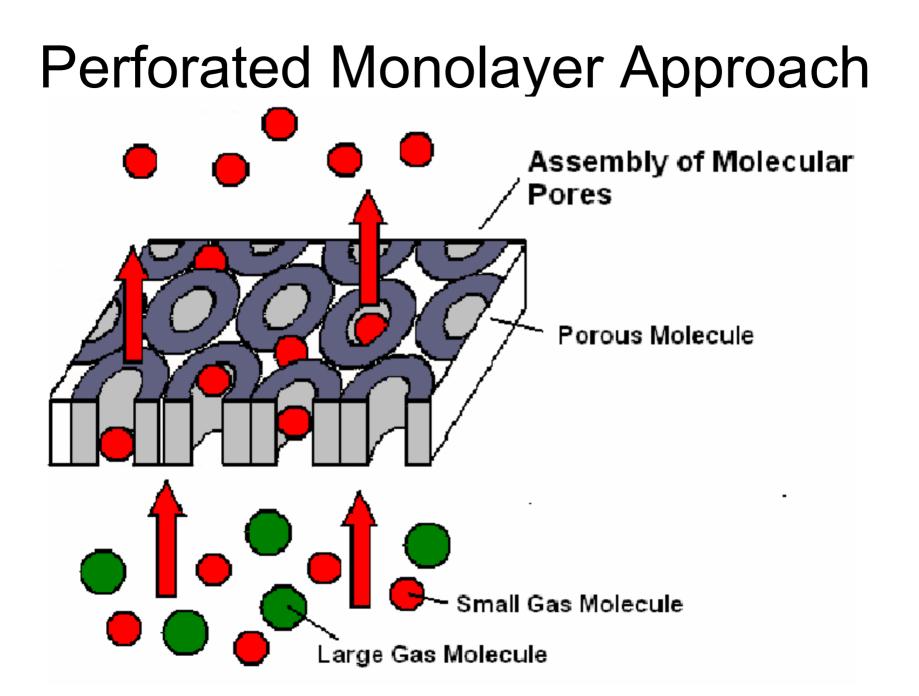


Protolyte A700

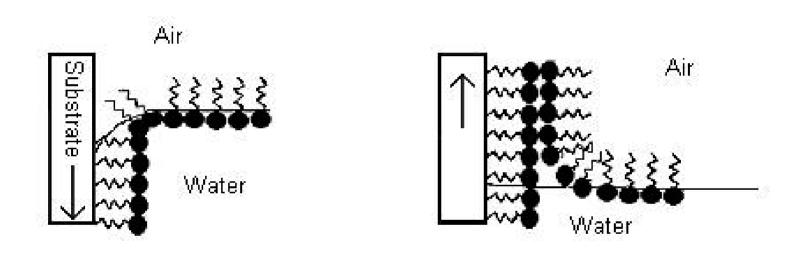


Summary – Ion Implantation

- Medium-energy ion beam treatment is a promising technique for developing barrier membranes selectively permeable to water vapor.
- Theoretical calculations are a useful adjunct for optimizing treatment conditions.
- XPS measurements of the surface reveal that ion bombardment leads to loss of fluorine, with the eventual formation of a carbonized layer.
- This two-pronged approach will ultimately yield a powerful technique for the development of permselective membranes for CB protective garments.

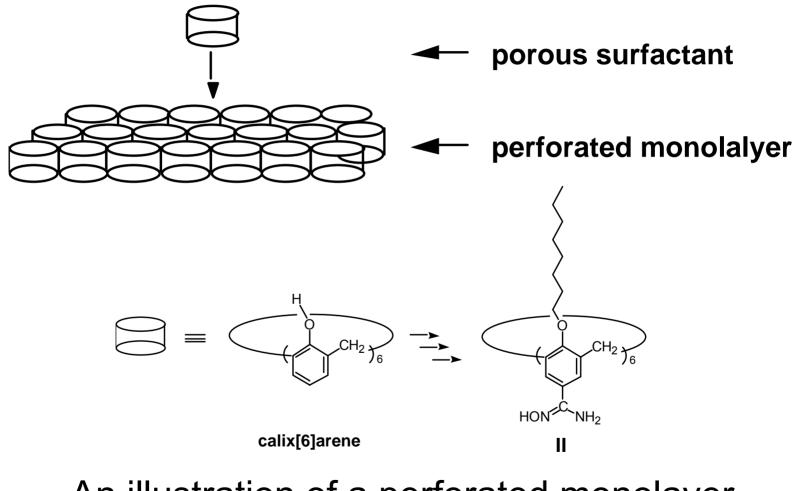


The Langmuir-Blodgett Method



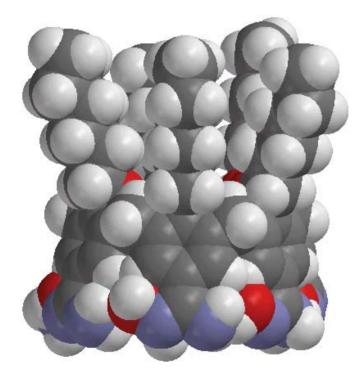
A stylized illustration showing a single surfactant monolayer being transferred to a hydrophobic support on a down-trip, followed by the transfer of a second monolayer on the up-trip, to form a bilayer.

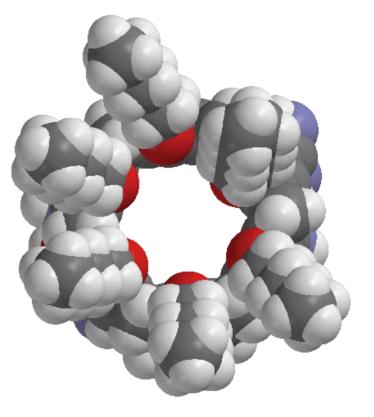
Perforated Monolayers



An illustration of a perforated monolayer formed from a porous surfactant.

. Space filling models of an analog of **II**

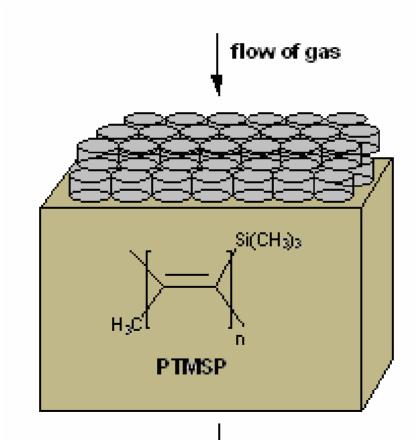




Side View

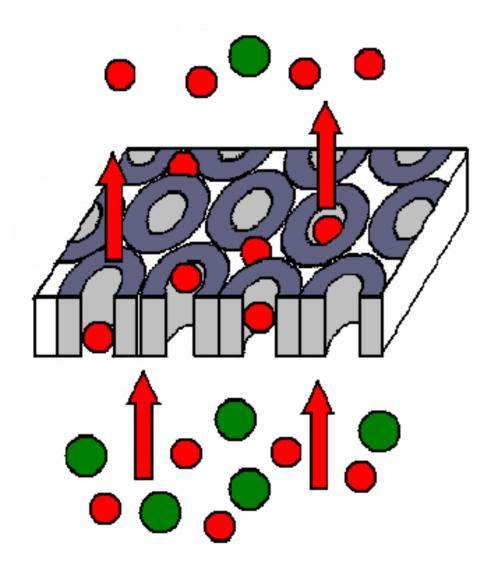
Top View

Composite membrane formed from a bilayer of **II** and poly[1-(trimethylsilyl)-1-propyne] (PTMSP)



Perforated Monolayer of II

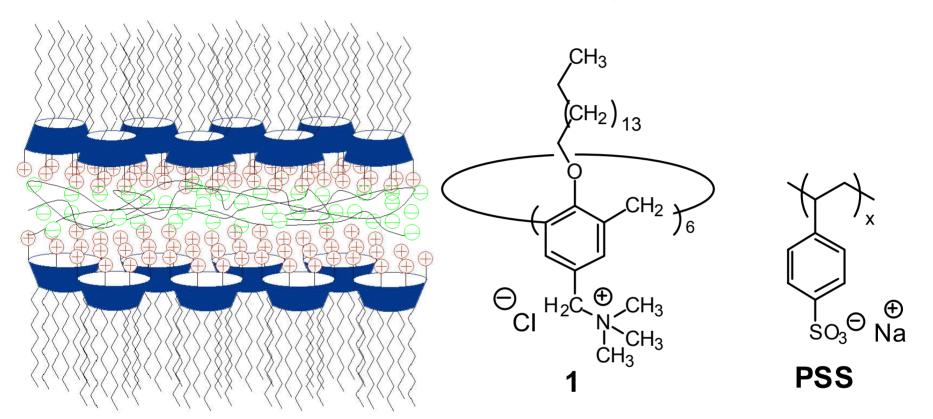
 $\alpha_{\text{He/N2}}$ = 18



Improved Perforated Monolayers through "Gluing"

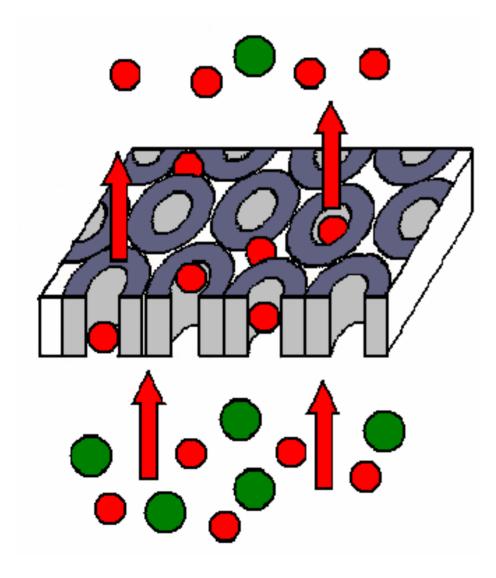
- Ionic cross-linking of a cationic calix[6]arene-based LB film by use of a water-soluble polyanion, would produce a two-dimensional network with enhanced stability
- Filling in void space (defects), the polymeric counterion would result in enhanced permeation selectivity

Glued LB Bilayer



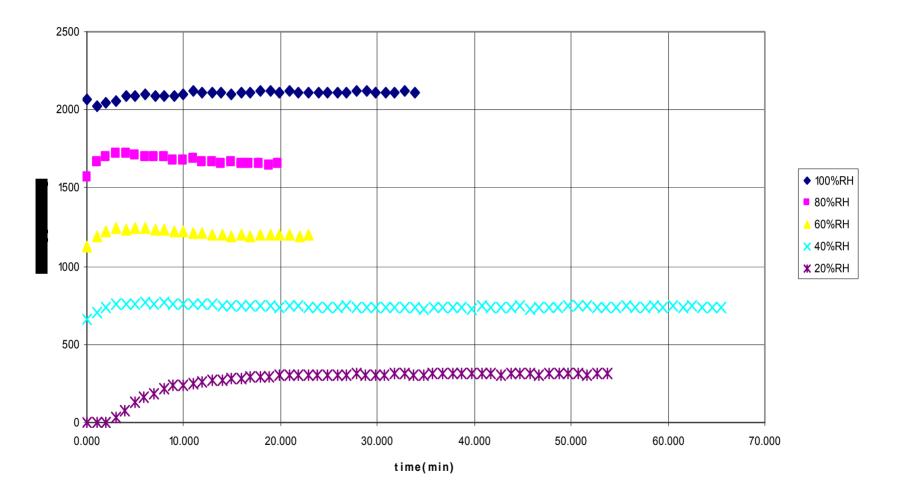
An illustration of a LB bilayer, made from a multiply charged calix[6]arene that has been glued together through the use of a polymeric counterion.

Perforated Monolayer of III & PSS

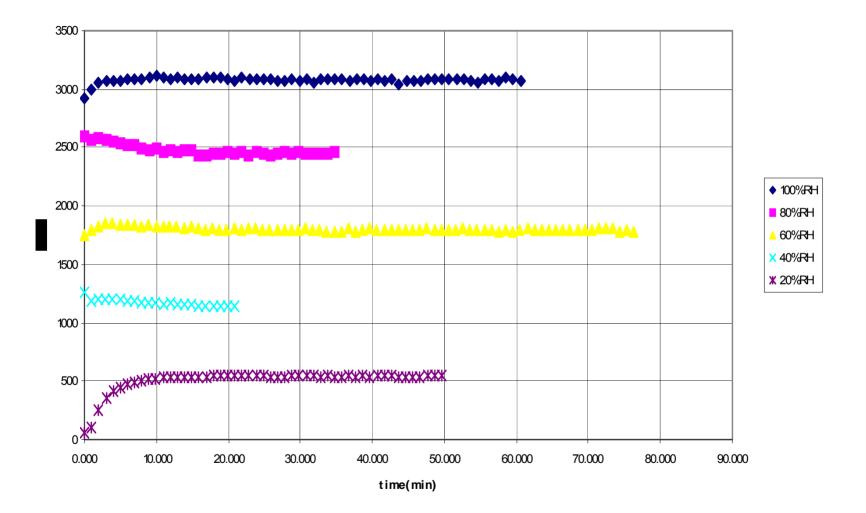


 $\alpha_{\text{He/N2}}$ = 240

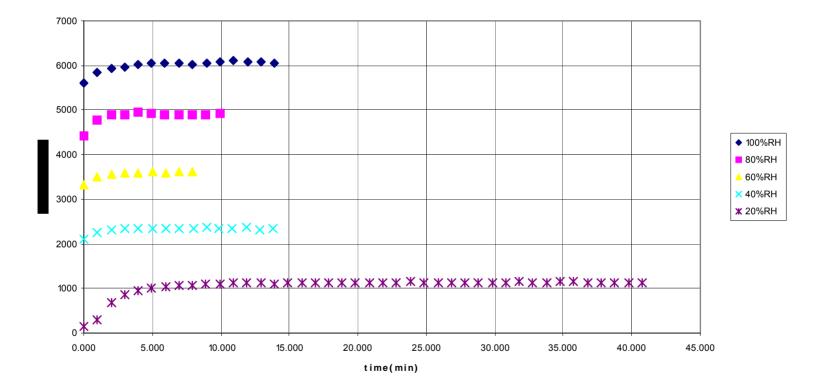
PDMS/PS WZ04013ABCDE



PTMSP WZ04023ABCDE



microporous PTFE



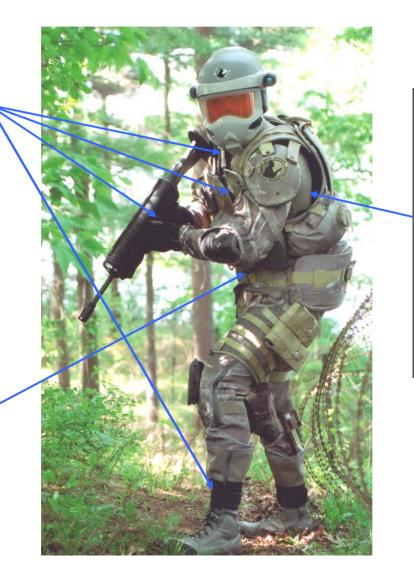
Summary – Monolayers

- The surface modification of organic polymers by a tightly packed monolayer of calix[6]arenes or other surfactants could constitute an attractive, selectively permeable barrier, allowing the passage of water vapor (perspiration), while serving as a barrier to chemical warfare agents
- Due to its ultrathin and microporous structure, it is expected that the flux of water across such a membrane would be maximized
- The composite membranes could be used in the protective layer of the next generation of chemical protective clothing
- These novel clothing ensembles would potentially be dramatically lighter weight than current systems

Summary/Challenges Clothing Operational Context

Improved system integration with suit, mask, helmet, gloves, boots, body armor, weapons, etc. (JSLIST Upgrade)

Reactive clothing materials with increased protection, reduced doffing hazard, and reduced logistics burden. (JSLIST Upgrade)



Cool, lightweight CB duty uniform based on nanofiber or membrane technology with increased mission duration and a reduced logistics burden. (JSLIST Upgrade)